X-Ray Luminescense of Thermoluminophors

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Abstract – Some of wide used thermoluminescent detectors of ionizing radiation are produced as polycrystals. Therefore investigation of nature of the luminescence and trapping centres, so as mechanism of thermoluminescence is complicate. Magnesium borate doped with dysprosium (MgB₄O₇:Dy) is widely used as thermoluminophor for personal dosimetry of gamma and beta radiation due to high sensitivity and close tissueequivalence. Now it is poorly investigated. X-ray luminescence in this phosphor was studied. MgB₄O₇:Tm was used as analog by chemical composition and CaSO₄:Dy (much more investigated than borates), as analog by dopant.

In the case of recombination kinetics for thermoluminophors, intensity of X-ray luminescence has to increase during the time of irradiation because of decrease of competition from filled traps and corresponding increase of number of holes captured by luminescence centers creating centers of recombination. Decrease of X-ray luminescence during the irradiation evidences about creation of local electron – hole complexes, which are excluded from recombination kinetics and appear in thermoluminescence only. For the set of samples investigated we have both types of kinetics. Schemes of the processes governed thermoluminescence and models of the trapping and luminescence centres in the phosphors investigated have been suggested.

1. Introduction

Magnesium borate doped with dysprosium $(MgB_4O_7:Dy)$ is widely used as thermoluminophor for personal dosimetry of gamma and beta radiation due to high sensitivity and close tissue-equivalence. Its analog $MgB_4O_7:Tm$ has more shortwave thermoluminescence (TL) (470 instead 580 nm), so it is more convenient for standard TL readers, but its efficiency is less than this of $MgB_4O_7:Dy$. Both the phosphors have main dosimetric TL peaks in the region of 200 °C.

Now we do not know for these phosphors satisfactory models of trapping and luminescence centers and processes occurred under ionizing irradiation and consequent release of the accumulated energy in the form of TL. It is connected with difficulties of preparation of the borate in single crystal form because of its incongruent melting and impossibility of growing the single crystals from its own melt. Therefore the processes occurred are to be studied without such informative tool as measuring of the optical absorption spectra, which is impossible in the case of powder samples available.

Any data on spectra of excitation or stimulation of the dopant emission for MgB₄O₇:Dy are absent in the literature available. Our own attempts to measure these spectra were not successful because of weak oscillator strength for f-f electron transitions, whereas f-d transitions falling in vacuum ultraviolet region were beyond our possibility. So, a most appropriate way to get some information about the phosphor was to measure spectra of X-ray luminescence (XL) in spite of the fact that X-ray irradiation changes initial electronic state of the phosphor. We intended to reveal influence of chemical composition, dopant, preliminary irradiation and synthesis conditions on the XL spectra. MgB₄O₇:Tm was used as analog by chemical composition and CaSO₄:Dy (much more investigated than borates), as analog by dopant.

Theoretical

The model consisting of two trap levels and one level of recombination centre (Fig. 1) is widely used in literature for simulating the processes occurring in crystals under irradiation [1]. Supralinearity, which appears in this scheme then the trap (2) is saturated is caused by competition of the traps. Corresponding set of equation is as follows:

$$dn_1/dt = A_1 n_e(N_1 - n_1)$$
(1)

$$dn_2/dt = A_2 n_e(N_2 - n_2)$$
⁽²⁾

$$dn_r/dt = -A_r n_e n_r + A_h n_h (N_r - n_r)$$
(3)

$$dn_e/dt = -A_1n_e(N_1 - n_1) - A_2n_e(N_2 - n_2) - A_rn_en_r$$
 (4)

$$dn_h/dt = A_h n_h (N_r - n_r)$$
(5)





Fig. 1. Electron energy levels scheme illustrated set of equations used for calculation of the dependences

capture by corresponding traps, its recombination and capture of hole by luminescence centre (creation of recombination centre), respectively; n_1 , n_2 , n_r , n_e , n_h are concentrations of electrons captured by corresponding traps, recombination centres, free electrons and holes, respectively; N_1 , N_2 , N_r are total concentrations of corresponding traps and luminescence centres, respectively.

Solution of this set of equations is time dependence of concentrations of corresponding centres. Usually in the case of gamma irradiation the crystal is considered as one great track permanently excited by radiation with intensity I_{ex} during time t. This situation can be described by the set of equations (1–5) with term I_{ex} *t added in the equations (4) and (5) to simulate permanent excitation.

The simplest case is excluding of the traps (2) from the recombination process. Then dose dependence of the filled traps (1) will be linear with saturation. Dependence for TL will be the same. Intensity of fotoluminescence (PL) will be proportional to concentration of luminescence centres $N_r - n_r$ and will decrease with increase of the recombination centres concentration. Intensity of XL will be determined by relation of probabilities of recombination and trapping and will increase with increase of concentration of filled traps and centres of recombination. $I_{XL} = I_{ex} * A_r n_r/(A_r n_r + A_2(N_2-n_2))$. These dependences are shown in the Figure 2 in schematic form.



Fig. 2. X-Ray luminescence spectra of the samples

2. Experimental

The borate samples were prepared in our laboratory, MgB₄O₇:Dy (TLD-580) and LiF:Mg,Ti (TLD-400) were produced in Stavropol and CaSO₄:Dy in Tartu. LiF:Mg,Cu,P (GR-200) was commercial product.

Spectra of X-ray excited luminescence were measured with the MDR-2 monochromator. The spectra were not corrected on spectral sensitivity of the unit. Preliminary irradiation was carried out with X-rays (35kV, 20mA). Irradiation dose was about 10⁴ Gy.

3. X-Ray spectra of the samples

The XL spectra (Fig. 2) are very close to TL spectra for the phosphors doped with the same rare earth (RE) ions [2].

Their shape is not changed sufficiently after irradiation, except the long wave band of $CaSO_4:Dy$. As for intensity, it decreases sufficiently (about 10 times) after irradiation for $CaSO_4:Dy$, more slightly (about 2 times) decreases for MgB₄O₇:Dy and increases (about 2 times) for MgB₄O₇:Tm.

4. Discussion

Because of short spin lattice relaxation times, ESR signals of Dy^{3+} were not observable at room temperature, but we have enough information about its transformations, based on investigations of optical spectra split under crystal field [3]. Luminescence in visible region for Dy^{3+} ion is connected with transitions between Stark splitting components of electron levels ${}^{4}F_{9/2} - {}^{6}H_{15/2}$ (480nm, 20000 cm⁻¹) and ${}^{4}F_{9/2} - {}^{6}H_{13/2}$ (575 nm, 17800 cm⁻¹). For Tm³⁺ (470 nm for free ion) it is transition ${}^{1}G_{4} - {}^{3}H_{6}$

In visible region unirradiated phosphors are colorless because of weak oscillator strength of f-f transitions. Intraconfiguration transition of such type are forbidden as intercombination ($\Delta S \neq 0$) and by selection rule for parity. Both the exclusions are violated by weak admixture of foreign wave functions. Notice that these mechanisms do not influence practically on positions of energetic levels but strongly effect on probabilities of transitions between them.

Transition f-d.s are situated in VUV region at wave-lengths shorter 180 nm [4] (width of forbidden gap in CaSO₄:Dy is 8.2 eV), and were unavailable in present investigation. Dy^{3+} is known to transform to Dy²⁺ under intensive gamma irradiation. This transformation is accompanied by appearance of absorption in visible region and luminescence in IR region. Fig. 3 shows such absorption for CaSO₄:Dy [5]. As a first approximation the absorption spectrum of irradiated CaSO₄:Dy corresponds to absorption of Dy²⁺ connected with transitions 4f-5d,6s (bands 21000 and 7000 cm⁻¹) by general view and value of optical density. Naturally, the CaSO₄:Dy spectrum is more blurred due to polycrystalline structure of the samples and more low symmetry of surrounding. Similarity between the details of the absorption curves obtained after different doses of irradiation are evidence of the

fact that these features are real structural elements of the spectra and can be connected with f-f transitions in Dy^{2^+} ion from ${}^3M_{10}$ and 5S_2 to ${}^5I_{5,6,7,8}$ states (these transitions occupy spectral between Stark splitting components of electron levels ${}^4F_{9/2} - {}^6H_{15/2}$ (480nm, 20000 cm⁻¹) and ${}^4F_{9/2} - {}^6H_{13/2}$ (575 nm, 17800 cm⁻¹). For Tm³⁺ (470 nm for free ion) it is transition ${}^1G_4 - {}^3H_6$ In visible region unirradiated phosphors are colorless



Fig. 3. Optical absorption of CaSO4:Dy irradiated with doses from 103 (D) up to 105 (A) rad

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Nature of electron traps in all samples activated by RE is beyond question. As for hole traps, we could not find in the literature available any evidences on presence of common accepted model, even for $CaSO_4:Dy$, not to mention about MgB₄O₇:Dy.

Comparison of Dy^{3+} spectra in CaSO₄:Dy and MgB₄O₇:Dy shows that for MgB₄O₇ in energetic spectrum of Dy^{3+} the level ${}^{4}F_{9/2}$ is split, and for CaSO₄:Dy it is level ${}^{6}H_{13/2}$.

Before discussing the influence of the samples structure on the spectra we have to stay on features of borate and sulphate structures. If to examine consequence of tetrahedral anionic complexes SO_4^{2-} , PO_4^{3-} , SiO_4^{4-} , BO_4^{5-} we should highlight such features as increase of the size and charge and decrease of internal bonds strengths (from 1.5 for SO_4^{2-} down to 0.75 for BO_4^{5-}) and corresponding decrease of rigidity of the structure. In particular, boron ion in borates can be not only in tetrahedral surrounding, but in trigonal too. In this case more freedom in ways of compensation of extra charge is possible.

In the case of recombination kinetics for thermoluminophors intensity of XL has to increase during the time of irradiation because of decrease of competition from filled traps and corresponding increase of holes captured by activator centers and created centers of recombination. Decrease of XL during the irradiation evidences about creation of local electron-hole complexes, which are excluded from recombination kinetics and appear in TL only. For the set of samples investigated we have both types of kinetics.

In the case of "normal" kinetics we see in XL spectra of irradiated samples the same centers but effected to influence of crystal field slightly changed because of presence in some distances of sufficient number of filled traps (electron and hole). In another case in the XL spectra of irradiated samples we see the centers which can not create complex because of their "abnormality" i.e. wrong way of local compensation. These centers were in the spectra from the very beginning but became visible only after excluding much more numerous "normal" centers. For such thermoluminophor dose dependence has to be linear with saturation because of reaching the limit of activator (saturation of dose dependence for CaSO₄:Dy at 10⁴ Gy corresponds to saturation of activator absorption). Many models of centers are discussed for CaSO₄:Dy. But, taking into account that after annealing up to 400 C only SO3⁻ ESR signal remains, the model of $Dv^{\bar{3}+}V_{Ca}SO_{3}^{-}$ center can be proposed. Double charged cation vacancy compensate two positive charges.

During irradiation Dy^{3+} consequently captures electron and it recombines with a hole, generating

luminescence. In opposite case free hole moving along anion rows meet double negative charged cation vacancy and localize near creating SO4- centre. One of positive charges becomes decompensate. But only Dy^{3+} can capture free electron because it moves along cation positions. After that the centre becomes compensated again but now it does not contain such great local charges as in previous case, so it is not so attractive for free carriers and can return to initial state only by thermally stimulated process. Different peaks of TL can be produced by this process depending on configuration of initial center or by including in the process other centers, such as O_2^{3-} (also indicated by ESR). Addition of Na⁺ as compensator of valency leads to stabilization of this center, and to quenching of TL. In the case of MgB₄O₇:Dy addition of Na⁺ (or Li⁺) as coactivator leads to increase of the TL output.

Investigation of XL sensitivity dependence for LiF:Mg,Cu,P (GR-200) on X-ray irradiation does not show significant changes up to dose close to saturation of TL output. Now we do not know how to explain such behavior.

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